

メタクリル酸ブチルのラットを用いた吸入によるがん原性試験報告書

試験番号：0849

APPENDICES

APPENDICES

- APPENDIX 1-1 IDENTITY OF BUTYL METHACRYLATE IN THE 2-YEAR
INHALATION STUDY
- APPENDIX 1-2 STABILITY OF BUTYL METHACRYLATE IN THE 2-YEAR
INHALATION STUDY
- APPENDIX 2 ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER
IN THE 2-YEAR INHALATION STUDY OF BUTYL
METHACRYLATE
- APPENDIX 3 METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY
AND BIOCHEMISTRY IN THE 2-YEAR INHALATION STUDY OF
BUTYL METHACRYLATE

APPENDIX 1-1

IDENTITY OF BUTYL METHACRYLATE IN THE 2-YEAR INHALATION STUDY

IDENTITY OF BUTYL METHACRYLATE IN THE 2-YEAR INHALATION STUDY

Test Substance : Butyl methacrylate (Wako Pure Chemical Industries, Ltd.)

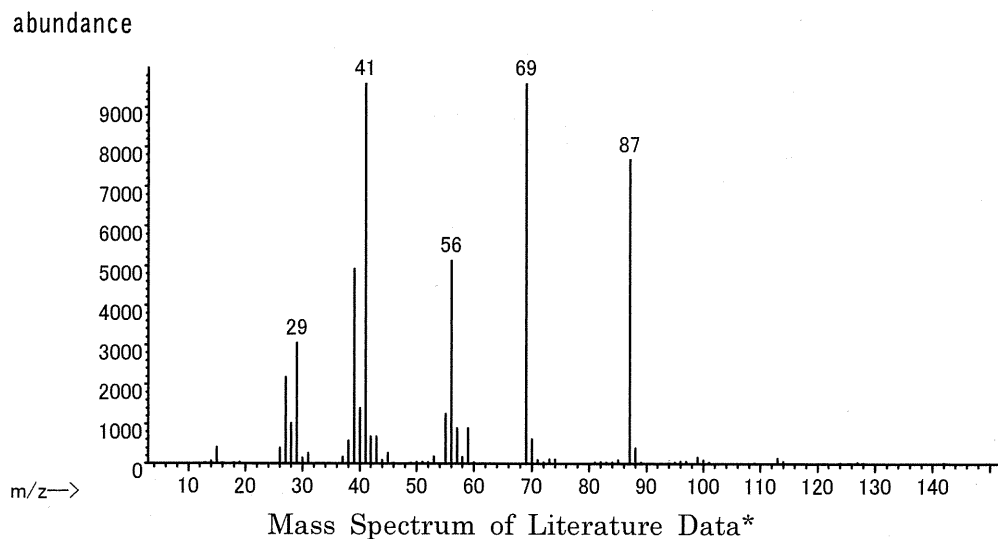
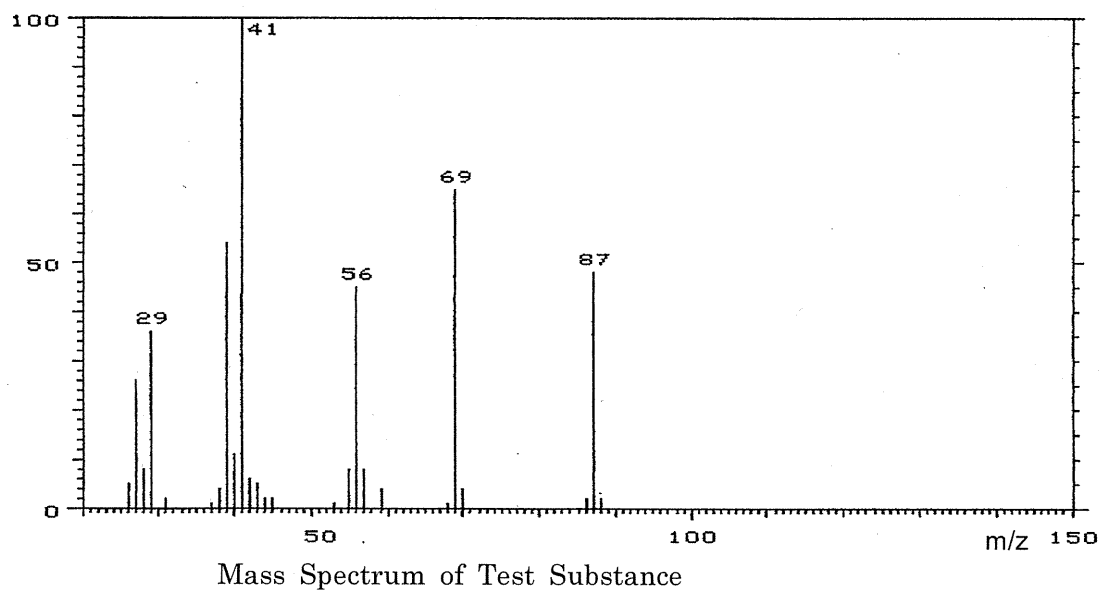
A. Lot No. : KPE0872

1. Mass Spectrometry

Instrument : Hitachi M-80B Mass Spectrometer

Ionization : EI (Electron Ionization)

Ionization Voltage : 70eV



Result: The mass spectrum was consistent with literature spectrum.

(*McLafferty FW, ed. 1994. Wiley Registry of Mass Spectral Data. 6th ed. New York, NY:John Wiley and Sons.)

2. Conclusion: The test substance was identified as butyl methacrylate by mass spectrum.

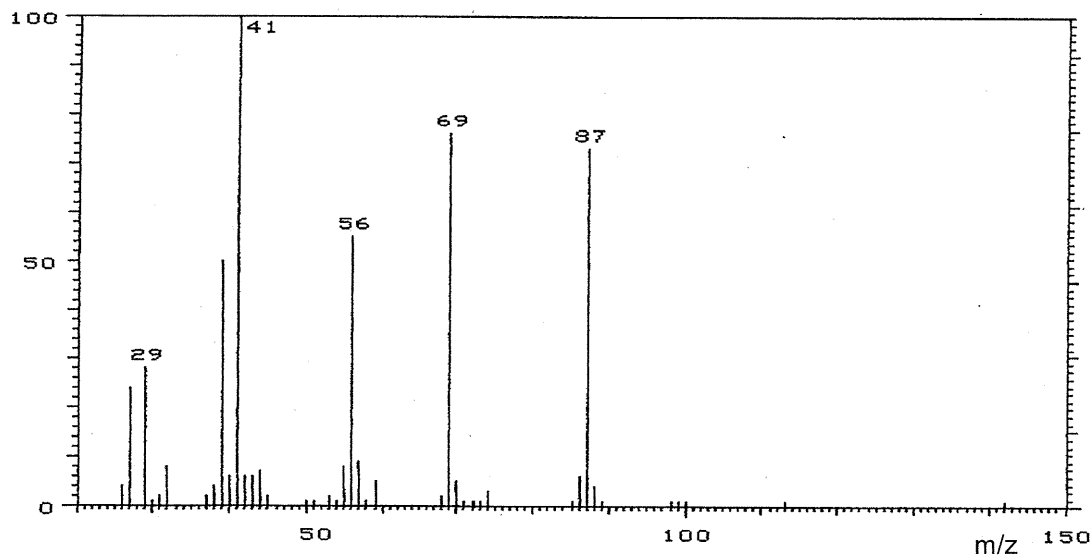
B. Lot No. : ECF5588

1. Mass Spectrometry

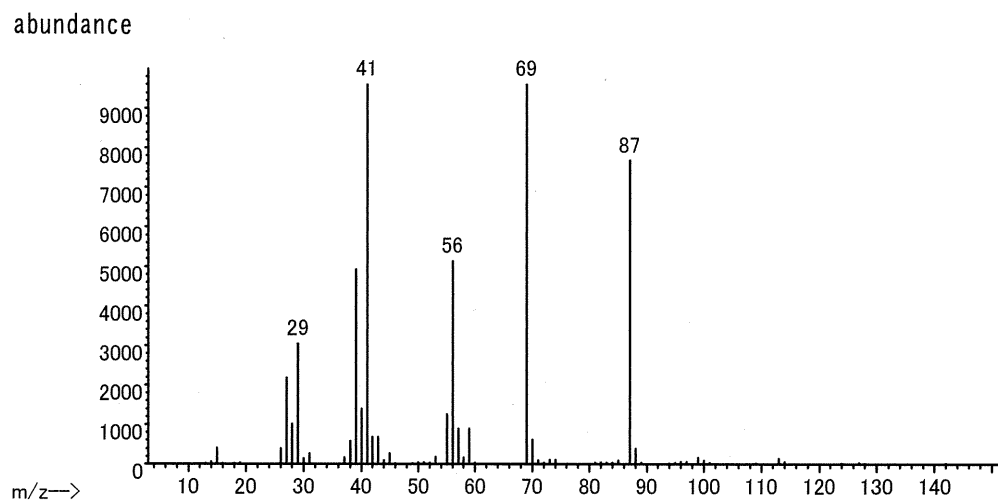
Instrument : Hitachi M-80B Mass Spectrometer

Ionization : EI (Electron Ionization)

Ionization Voltage : 70eV



Mass Spectrum of Test Substance



Mass Spectrum of Literature Data*

Result: The mass spectrum was consistent with literature spectrum.

(*McLafferty FW, ed. 1994. Wiley Registry of Mass Spectral Data. 6th ed.
New York, NY:John Wiley and Sons.)

2. Conclusion: The test substance was identified as butyl methacrylate by mass spectrum.

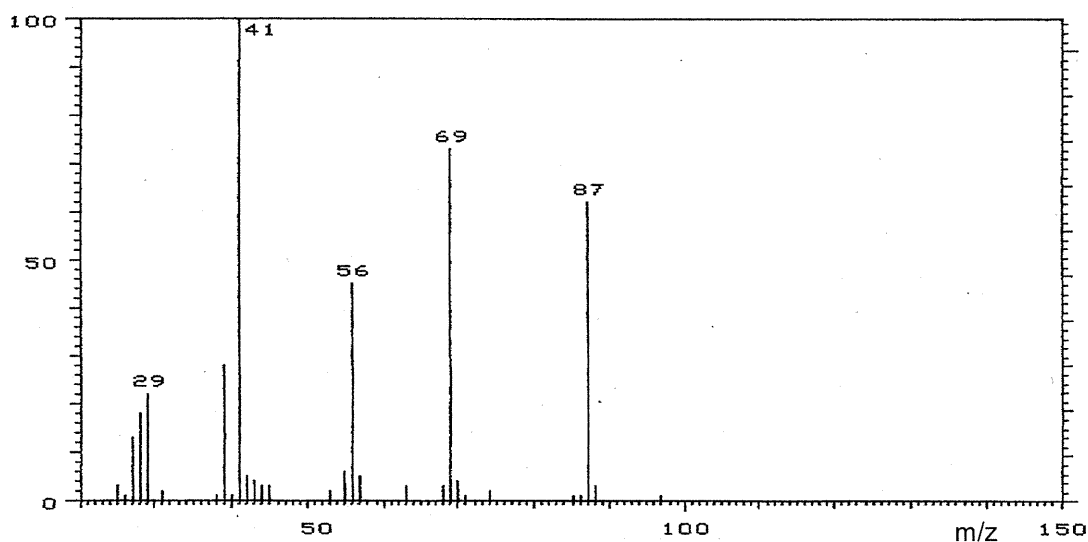
C. Lot No. : DSK6422

1. Mass Spectrometry

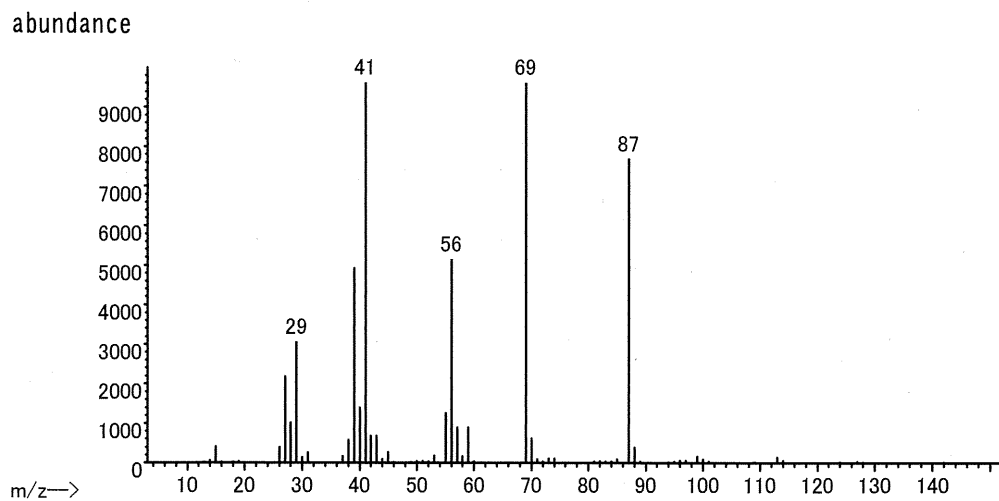
Instrument : Hitachi M-80B Mass Spectrometer

Ionization : EI (Electron Ionization)

Ionization Voltage : 70eV



Mass Spectrum of Test Substance



Mass Spectrum of Literature Data*

Result: The mass spectrum was consistent with literature spectrum.

(*McLafferty FW, ed. 1994. Wiley Registry of Mass Spectral Data. 6th ed.
New York, NY:John Wiley and Sons.)

2. Conclusion: The test substance was identified as butyl methacrylate by mass spectrum.

APPENDIX 1-2

STABILITY OF BUTYL METHACRYLATE IN THE 2-YEAR INHALATION STUDY

STABILITY OF BUTYL METHACRYLATE IN THE 2-YEAR INHALATION STUDY

Test Substance : Butyl methacrylate (Wako Pure Chemical Industries, Ltd.)

A. Lot No. : KPE0872

1. Gas Chromatography

Instrument : Agilent Technologies 5890A Gas Chromatograph

Column : INNOWAX (0.53 mm ϕ $\times$ 60 m)

Column Temperature : 140 $^{\circ}$ C

Flow Rate : 20 mL/min

Detector : FID (Flame Ionization Detector)

Injection Volume : 1 μ L

Date Analyzed	Peak No.	Retention Time (min)	Area (%)
2015.01.26	1	1.608	100
2016.03.07	1	1.582	100

Result: Gas chromatography indicated one major peak (peak No.1) analyzed on 2015.1.26 and one major peak (peak No.1) analyzed on 2016.3.7. No new trace impurity peak in the test substance analyzed on 2016.3.7 was detected.

2. Conclusion: The test substance was stable for the period that the test substance had been used for the study.

B. Lot No. : ECF5588

1. Gas Chromatography

Instrument : Agilent Technologies 5890A Gas Chromatograph

Column : INNOWAX (0.53 mm ϕ $\times$ 60 m)

Column Temperature : 140 °C

Flow Rate : 20 mL/min

Detector : FID (Flame Ionization Detector)

Injection Volume : 1 μ L

Date Analyzed	Peak No.	Retention Time (min)	Area (%)
2016.03.02	1	1.575	100
2016.10.27	1	1.585	100

Result: Gas chromatography indicated one major peak (peak No.1) analyzed on 2016.3.2 and one major peak (peak No.1) analyzed on 2016.10.27. No new trace impurity peak in the test substance analyzed on 2016.10.27 was detected.

2. Conclusion: The test substance was stable for the period that the test substance had been used for the study.

C. Lot No. : DSK6422

1. Gas Chromatography

Instrument : Agilent Technologies 5890A Gas Chromatograph

Column : INNOWAX (0.53 mm ϕ $\times$ 60 m)

Column Temperature : 140 °C

Flow Rate : 20 mL/min

Detector : FID (Flame Ionization Detector)

Injection Volume : 1 μ L

Date Analyzed	Peak No.	Retention Time (min)	Area (%)
2016.10.25	1	1.585	100
2017.02.15	1	1.582	100

Result: Gas chromatography indicated one major peak (peak No.1) analyzed on 2016.10.25 and one major peak (peak No.1) analyzed on 2017.2.15. No new trace impurity peak in the test substance analyzed on 2017.2.15 was detected.

2. Conclusion: The test substance was stable for the period that the test substance had been used for the study.

APPENDIX 2

ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 2-YEAR INHALATION STUDY OF BUTYL METHACRYLATE

ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 2-YEAR
INHALATION STUDY OF BUTYL METHACRYLATE

Group Name	Temperature (°C)	Humidity (%)	Ventilation Rate (L/min)		Air Change (time/h)	
	Mean ± S.D.	Mean ± S.D.	Mean ± S.D.* ¹	Mean ± S.D.* ²	Mean* ¹	Mean* ²
Control	23.1 ± 0.1	52.6 ± 2.6	768.3 ± 6.5	1520.7 ± 14.5	6.1	12.0
30 ppm	23.1 ± 0.1	52.0 ± 2.6	768.0 ± 6.6	1524.8 ± 13.8	6.1	12.0
125 ppm	23.2 ± 0.2	53.9 ± 2.7	770.6 ± 6.0	1523.9 ± 14.3	6.1	12.0
500 ppm	23.2 ± 0.2	50.2 ± 2.6	768.4 ± 5.5	1524.4 ± 9.9	6.1	12.0

* 1: Exposure period

* 2: After exposure period

APPENDIX 3

METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY
AND BIOCHEMISTRY IN THE 2-YEAR INHALATION STUDY OF
BUTYL METHACRYLATE

METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY AND BIOCHEMISTRY IN THE 2-YEAR INHALATION STUDY OF BUTYL METHACRYLATE

Item	Method	Unit	Decimal place
Hematology			
Red blood cell (RBC)	Light scattering method ¹⁾	$\times 10^6/\mu\text{L}$	2
Hemoglobin(Hgb)	Cyanmethemoglobin method ¹⁾	g/dL	1
Hematocrit(Hct)	Calculated as $\text{RBC} \times \text{MCV} / 10$ ¹⁾	%	1
Mean corpuscular volume(MCV)	Light scattering method ¹⁾	fL	1
Mean corpuscular hemoglobin(MCH)	Calculated as $\text{Hgb} / \text{RBC} \times 10$ ¹⁾	pg	1
Mean corpuscular hemoglobin concentration (MCHC)	Calculated as $\text{Hgb} / \text{Hct} \times 100$ ¹⁾	g/dL	1
Platelet	Light scattering method ¹⁾	$\times 10^3/\mu\text{L}$	0
Reticulocyte	Light scattering method ¹⁾	%	1
White blood cell(WBC)	Light scattering method ¹⁾	$\times 10^3/\mu\text{L}$	2
Differential WBC	Light scattering method ¹⁾	%	0
Biochemistry			
Total protein(TP)	Biuret method ²⁾	g/dL	1
Albumin (Alb)	BCG method ²⁾	g/dL	1
A/G ratio	Calculated as $\text{Alb} / (\text{TP} - \text{Alb})$ ²⁾	—	1
T-bilirubin	BOD method ²⁾	mg/dL	2
Glucose	GlcK·G-6-PDH method ²⁾	mg/dL	0
T-cholesterol	CE·COD·POD method ²⁾	mg/dL	0
Triglyceride	MGLP·GK·GPO·POD method ²⁾	mg/dL	0
Phospholipid	PLD·ChOD·POD method ²⁾	mg/dL	0
Aspartate aminotransferase (AST)	JSCC method ²⁾	U/L	0
Alanine aminotransferase (ALT)	JSCC method ²⁾	U/L	0
Lactate dehydrogenase (LDH)	JSCC method ²⁾	U/L	0
Alkaline phosphatase (ALP)	JSCC method ²⁾	U/L	0
γ -Glutamyl transpeptidase (γ -GTP)	JSCC method ²⁾	U/L	1
Creatine kinase (CK)	JSCC method ²⁾	U/L	0
Urea nitrogen	Urease·GLDH method ²⁾	mg/dL	1
Creatinine	Creatinase·SOD·POD method ²⁾	mg/dL	2
Sodium	Ion selective electrode method ²⁾	mEq/L	0
Potassium	Ion selective electrode method ²⁾	mEq/L	1
Chloride	Ion selective electrode method ²⁾	mEq/L	0
Calcium	OCPC method ²⁾	mg/dL	1
Inorganic phosphorus	PNP·XOD·POD method ²⁾	mg/dL	1

1) Automatic blood cell analyzer (ADVIA120 : Siemens Healthcare Diagnostics Inc.)

2) Automatic analyzer (Hitachi 7080 : Hitachi,Ltd.)